

Synthesis of Polycyclic Molecules by Double C(sp²)–H/C(sp³)–H Arylations with a Single Palladium Catalyst

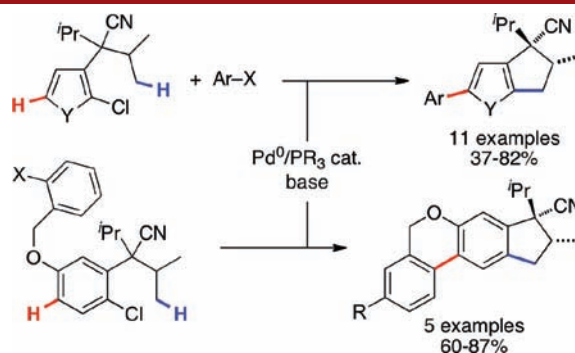
Cathleen Pierre and Olivier Baudoin*

Université Claude Bernard Lyon 1, CNRS UMR 5246 - Institut de Chimie et Biochimie Moléculaires et Supramoléculaires, CPE Lyon, 43 Boulevard du 11 Novembre 1918, 69622 Villeurbanne, France

olivier.baudoin@univ-lyon1.fr

Received February 3, 2011

ABSTRACT



Polycyclic molecules were obtained in good yields by double C(sp²)–H/C(sp³)–H aryations mediated by a single palladium/phosphine catalyst. Both double intermolecular/intramolecular and intramolecular/intramolecular C–C couplings were performed successfully, which indicates that this concept has a broad applicability for the rapid construction of molecular complexity.

In recent years, transition-metal-catalyzed C–H bond functionalization has emerged as a powerful tool to transform otherwise unreactive C–H bonds into carbon–carbon or carbon–heteroatom bonds.^{1,2} In particular, C–H arylation has become an attractive alternative to traditional C–C cross-coupling reactions due to the

minimization of stoichiometric metallic waste and the costs associated with the preparation of starting materials.³ Efforts by our group⁴ and others^{5,6} in the context of alkane C(sp³)–H arylation under palladium(0) catalysis have led to a greater scope for intramolecular reactions, generating

(1) *Handbook of C–H Transformations*; Dyker, G., Ed.; Wiley-VCH: Weinheim, 2005.

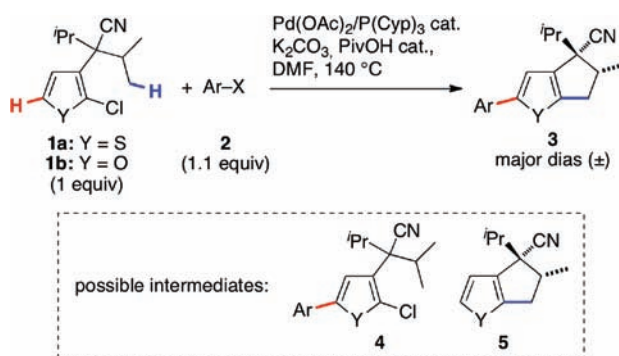
(2) *Topics in Current Chemistry*, Vol. 292; Yu, J.-Q., Shi, Z., Eds.; Springer: Heidelberg, 2010.

(3) Selected recent reviews: (a) Campeau, L.-C.; Fagnou, K. *Chem. Commun.* **2006**, 1253. (b) Alberico, D.; Scott, M. E.; Lautens, M. *Chem. Rev.* **2007**, *107*, 174. (c) McGlacken, G. P.; Bateman, L. M. *Chem. Soc. Rev.* **2009**, *38*, 2447. (d) Bellina, F.; Rossi, R. *Tetrahedron* **2009**, *65*, 10269. (e) Ackermann, L.; Vicente, R.; Kapdi, A. R. *Angew. Chem., Int. Ed.* **2009**, *48*, 9792.

(4) (a) Baudoin, O.; Herrbach, A.; Guéritte, F. *Angew. Chem., Int. Ed.* **2003**, *42*, 5736. (b) Hitce, J.; Retailleau, P.; Baudoin, O. *Chem.—Eur. J.* **2007**, *13*, 792. (c) Hitce, J.; Baudoin, O. *Adv. Synth. Catal.* **2007**, *349*, 2054. (d) Chaumontet, M.; Piccardi, R.; Audic, N.; Hitce, J.; Peglion, J.-L.; Clot, E.; Baudoin, O. *J. Am. Chem. Soc.* **2008**, *130*, 15157. (e) Chaumontet, M.; Piccardi, R.; Baudoin, O. *Angew. Chem., Int. Ed.* **2009**, *48*, 179. (f) Rousseaux, S.; Davi, M.; Sofack-Kreutzer, J.; Pierre, C.; Kefalidis, C. E.; Clot, E.; Fagnou, K.; Baudoin, O. *J. Am. Chem. Soc.* **2010**, *132*, 10706.

(5) Selected work on intramolecular C(sp³)–H arylation: (a) Dong, C.-G.; Hu, Q.-S. *Angew. Chem., Int. Ed.* **2006**, *45*, 2289. (b) Ren, H.; Knochel, P. *Angew. Chem., Int. Ed.* **2006**, *45*, 3462. (c) Ren, H.; Li, Z.; Knochel, P. *Chem.—Asian J.* **2007**, *2*, 416. (d) Lafrance, M.; Gorelsky, S. I.; Fagnou, K. *J. Am. Chem. Soc.* **2007**, *129*, 14570. (e) Watanabe, T.; Oishi, S.; Fujii, N.; Ohno, H. *Org. Lett.* **2008**, *10*, 1759. (f) Salcedo, A.; Neuville, L.; Zhu, J. *J. Org. Chem.* **2008**, *73*, 3600. (g) Hsiao, C.-C.; Lin, Y.-K.; Liu, C.-J.; Wu, T.-C.; Wu, Y.-T. *Adv. Synth. Catal.* **2010**, *352*, 3267. (h) Rousseaux, S.; Gorelsky, S. I.; Chung, B. K. W.; Fagnou, K. *J. Am. Chem. Soc.* **2010**, *132*, 10692.

(6) Selected work on intermolecular C(sp³)–H arylation: (a) Barder, T. E.; Walker, S. D.; Martinelli, J. R.; Buchwald, S. L. *J. Am. Chem. Soc.* **2005**, *127*, 4685. (b) Niwa, T.; Yorimitsu, H.; Oshima, K. *Org. Lett.* **2007**, *9*, 2373. (c) Campeau, L.-C.; Schipper, D. J.; Fagnou, K. *J. Am. Chem. Soc.* **2008**, *130*, 3266. (d) Mousseau, J. J.; Larivée, A.; Charette, A. B. *Org. Lett.* **2008**, *10*, 1641. (e) Wasa, M.; Engle, K. M.; Yu, J.-Q. *J. Am. Chem. Soc.* **2009**, *131*, 9886. (f) Burton, P. M.; Morris, J. A. *Org. Lett.* **2010**, *12*, 5359. (g) Renaudat, A.; Jean-Gérard, L.; Jazzar, R.; Kefalidis, C. E.; Clot, E.; Baudoin, O. *Angew. Chem., Int. Ed.* **2010**, *49*, 7261.

Table 1. Double Intermolecular C(sp²)-H/Intramolecular C(sp³)-H Arylation of Thiophenes and Furans^a

entry	X	product	yield (%) ^b	dr ^c	entry	X	product	yield (%) ^b	dr ^c
1	Br		37	>95:5	8	Cl		62	4:1
2	Br		82	16:1	9	Br		0 ^d	–
3	Cl		60	4:1	10	Br		65	7:1
4	Br		65	7:1	11	Br		78	6:1
5	Br		79	6:1	12	Br		50	5:1
6	Br		80	5:1					
7	Br		72	17:1					

^a Reaction conditions: heteroaryl chloride (1 equiv), aryl halide (1.1 equiv), Pd(OAc)₂ (5 mol %), P(Cyp)₃•HBF₄ (20 mol %), K₂CO₃ (2 equiv), PivOH (30 mol %), DMF, 140 °C, 6–24 h. ^b Yield of the isolated product. ^c Determined by ¹H NMR analysis of the crude reaction mixture. ^d 1:1 ratio of 1a/4 after 14 h.

various motifs of synthetically useful fused carbocycles and heterocycles.⁷ We report herein that intramolecular

(7) Review: Jassar, R.; Hitce, J.; Renaudat, A.; Sofack-Kreutzer, J.; Baudoin, O. *Chem.—Eur. J.* **2010**, *16*, 2654.

(8) For recent examples of multiple C(sp²)-H arylations: (a) Huang, Q.; Fazio, A.; Dai, G.; Campo, M. A.; Larock, R. C. *J. Am. Chem. Soc.* **2004**, *126*, 7460. (b) Wakui, H.; Kawasaki, S.; Satoh, T.; Miura, M.; Nomura, M. *J. Am. Chem. Soc.* **2004**, *126*, 8658. (c) Kamikawa, K.; Takemoto, I.; Takemoto, S.; Matsuzaka, H. *J. Org. Chem.* **2007**, *72*, 7406. (d) Ohno, H.; Iuchi, M.; Fujii, N.; Tanaka, T. *Org. Lett.* **2007**, *9*, 4813. (e) Majumdar, K. C.; Chakravorty, S.; De, N. *Tetrahedron Lett.* **2008**, *49*, 3419. (f) Join, B.; Yamamoto, T.; Itami, K. *Angew. Chem., Int. Ed.* **2009**, *48*, 3644. (g) Della Ca', N.; Maestri, G.; Catellani, M. *Chem.—Eur. J.* **2009**, *15*, 7850. (h) Thirunavukkarasu, V. S.; Parthasarathy, K.; Cheng, C.-H. *Chem.—Eur. J.* **2010**, *16*, 1436. (i) Shibahara, F.; Yamaguchi, E.; Murai, T. *Chem. Commun.* **2010**, *46*, 2471.

C(sp³)-H arylations can be combined with inter- or intramolecular C-H arylations in the presence of the same palladium catalyst. These new types of double C-H arylations⁸ allow the rapid construction of molecular complexity and provide an efficient access to original polycyclic molecules.

The double intermolecular C(sp²)-H/intramolecular C(sp³)-H arylation of chlorothiophene 1a with aryl halides 2 was first investigated (Table 1). Individual C-H arylations at play in this reaction are known,^{3,4f,9} but we wondered if a single batch of catalyst was able to perform both operations in one pot without undergoing deactivation intermediately. This was observed to be the case, and

the best results were obtained using the Pd(OAc)₂/tricyclopentylphosphine [P(Cyp)₃] catalyst previously developed for intramolecular arylation,^{4f} in the presence of catalytic pivalic acid¹⁰ and K₂CO₃ in DMF at 140 °C. Remarkably, only 1.1 equiv of aryl halide was sufficient to achieve complete conversion under these conditions. The reaction proceeded more efficiently with aryl halides **2** bearing an *ortho* or *meta* electron-withdrawing or neutral group (entries 2–8), whereas other aryl bromides such as bromobenzene (entry 1) and *p*-bromotoluene (entry 9) gave mixtures of unreacted **1a**, mono-C(sp²)-H arylation (**4**), and bis-arylation (**3**) products.

Interestingly, intermolecular C(sp²)-H arylation occurred before intramolecular C(sp³)-H arylation with aryl bromides, as shown by the observation of compound **4** and the absence of compound **5** at incomplete substrate conversion. In addition, the diastereoselectivity observed for products **3** that were obtained in the presence of aryl bromides was usually higher (dr ≥ 5:1) than that observed for product **5** that was obtained in the absence of aryl bromide (dr = 4:1). When *o*-chlorotoluene was employed as the coupling partner (entry 3) instead of *o*-bromotoluene (entry 2), intramolecular C(sp³)-H arylation occurred first, as shown by the observation of compound **5** and the absence of compound **4** at incomplete substrate conversion. The 4:1 diastereoselectivity observed for **3b** in this case (entry 3) is also consistent with this analysis. With the studied aryl bromides **2**, oxidative addition of the C-Br bond to Pd seems to occur first and the Pd catalyst is thus committed first to the intermolecular C(sp²)-H arylation. In contrast, with *o*-chlorotoluene, oxidative addition seems to occur first from the more electron-deficient heteroarene C-Cl bond and thus intramolecular C(sp³)-H arylation is taking place first. Additional factors such as the accelerating effect of the heteroarene chlorine atom on the intermolecular C(sp²)-H arylation¹¹ and the difference in C(sp³)-H and C(sp²)-H bond strengths may also be at play in this double C-H arylation process. Interestingly, furan **1b** could be arylated successfully, similar to thiophene **1a** (entries 10–12), although longer reaction times were required due to the lower reactivity of the furan motif toward intermolecular C(sp²)-H arylation. Major diastereoisomers **3** obtained from this double arylation process all possessed the *cis* configuration, as determined by NOESY experiments and by X-ray diffraction analysis for compound **3f** (Figure 1). The latter revealed a highly planar biaryl structure, indicating that the difference in diastereoselectivities observed with the various aryl bromides arise from electronic factors rather than from steric ones.

A double intramolecular C(sp²)-H/C(sp³)-H arylation was next investigated, and dihalogenated substrates **6**, synthesized in five steps from commercially available

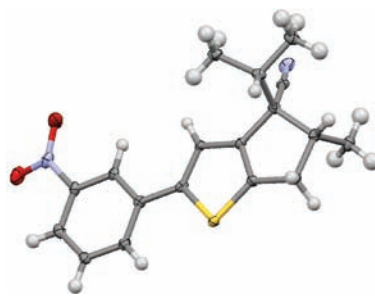
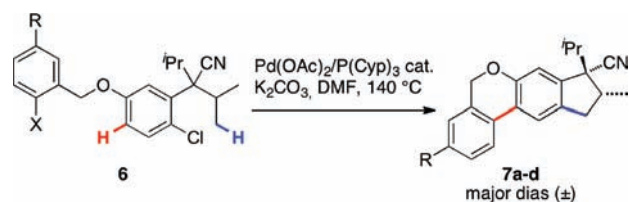


Figure 1. X-ray crystal structure of compound **3f** (30% probability ellipsoids plot).

compounds (see the Supporting Information), were studied as a proof of concept (Table 2). The same reaction conditions as those in the mixed inter/intramolecular arylation (Table 1) were employed, with the exception of pivalic acid which was not required to obtain optimal yields in this case.¹² Under these conditions, fused tetracyclic compounds **7a–d** were obtained in good yields and with comparable diastereoselectivities, regardless of the electron-withdrawing/-donating nature of the R substituent on the haloarene ring. Monitoring these reactions at incomplete conversion revealed that intramolecular C(sp²)-H arylation preceded intramolecular C(sp³)-H arylation with X = Br, which again results probably from the faster initial oxidative addition of the C-Br bond to the Pd catalyst. The same reactivity trend was observed with X = Cl (entry 2), for which the first oxidative addition probably occurs at the least sterically hindered C-Cl

Table 2. Double Intramolecular C(sp²)-H/C(sp³)-H Arylation^a



entry	X	R	product	yield (%) ^b	dr ^c
1	Br	H	7a	68	4.5:1
2	Cl	H	7a	83	4.5:1
3	Br	F	7b	72	5:1
4	Br	CF ₃	7c	87	5:1
5	Br	OMe	7d	60	5:1

^a Reaction conditions: Pd(OAc)₂ (5 mol %), P(Cyp)₃•HBF₄ (20 mol %), K₂CO₃ (2 equiv), DMF, 140 °C, 24–36 h. ^b Yield of the isolated product. ^c Determined by ¹H NMR analysis of the crude reaction mixture.

(12) For similar conditions developed for intramolecular C(sp²)-H arylations: (a) Campeau, L.-C.; Parisien, M.; Leblanc, M.; Fagnou, K. *J. Am. Chem. Soc.* **2004**, *126*, 9186. (b) Campeau, L.-C.; Parisien, M.; Jean, A.; Fagnou, K. *J. Am. Chem. Soc.* **2006**, *128*, 581.

(9) For similar conditions developed for intermolecular C-H arylations of heteroarenes: Liégault, B.; Lapointe, D.; Caron, L.; Vlassova, A.; Fagnou, K. *J. Org. Chem.* **2009**, *74*, 1826.

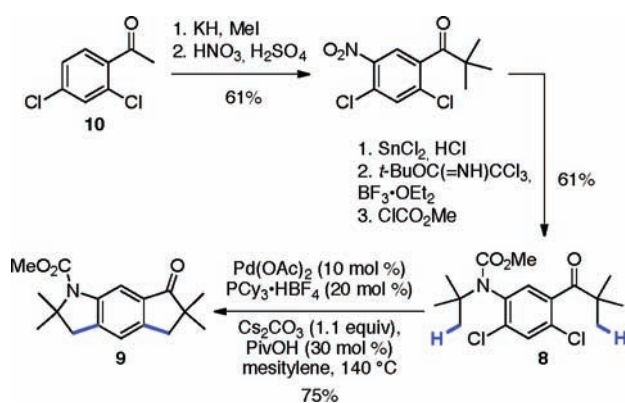
(10) Lafrance, M.; Fagnou, K. *J. Am. Chem. Soc.* **2006**, *128*, 16496.

(11) Liégault, B.; Petrov, I.; Gorelsky, S. I.; Fagnou, K. *J. Org. Chem.* **2010**, *75*, 1047.

bond. This order of arylation events may explain why a complete regioselectivity was observed for the intramolecular C(sp²)-H arylation, which occurred as expected at the most sterically accessible C(sp²)-H bond *para* to the quaternary benzylic carbon.¹²

Finally, we examined the feasibility of a double intramolecular C(sp³)-H/C(sp³)-H arylation under the same guidelines.^{5b,c} As in previous examples, the corresponding individual C-H arylations are known,^{4f} but we again wondered if a single batch of catalyst could perform both operations. To this purpose, we chose to study the reactivity of dichloride **8** (Scheme 1), which was synthesized in five steps, 37% overall yield, from 2,4-dichloroacetophenone

Scheme 1. Double Intramolecular C(sp³)-H/C(sp³)-H Arylation



10. We considered as well other dihalogenated substrates, but **8** turned out to be the most accessible. Gratifyingly, compound **8** furnished original fused tricyclic product **9** in 75% yield, under conditions previously optimized for individual C-H arylations, i.e. with tricyclohexylphosphine as the Pd ligand, Cs₂CO₃/pivalic acid as the base/additive, and mesitylene as the solvent.^{4f}

In conclusion, we have synthesized original polycyclic molecules in good yields and diastereoselectivities by double C-H arylations mediated by a single palladium/phosphine catalyst. Both double intermolecular/intramolecular and intramolecular/intramolecular C-C couplings were performed successfully, which indicates that this concept may be broadly used for the rapid construction of molecular complexity.

Acknowledgment. We thank Ministère de l'Enseignement Supérieur et de la Recherche (fellowship to C.P.), Agence Nationale de la Recherche, Institut Universitaire de France, and Johnson Matthey PLC (loan of palladium acetate). We are also grateful to E. Jeanneau, C. Toppan, and D. Bouchu (Université Claude Bernard Lyon 1) for X-ray diffraction, NMR, and MS analyses, respectively, and N. Martin for carefully reading the experimental part.

Supporting Information Available. Full characterization of all new compounds, detailed experimental procedures, copies of NMR spectra for target molecules, and X-ray crystal structure data (CIF) for compound **3f**. This material is available free of charge via the Internet at <http://pubs.acs.org>.